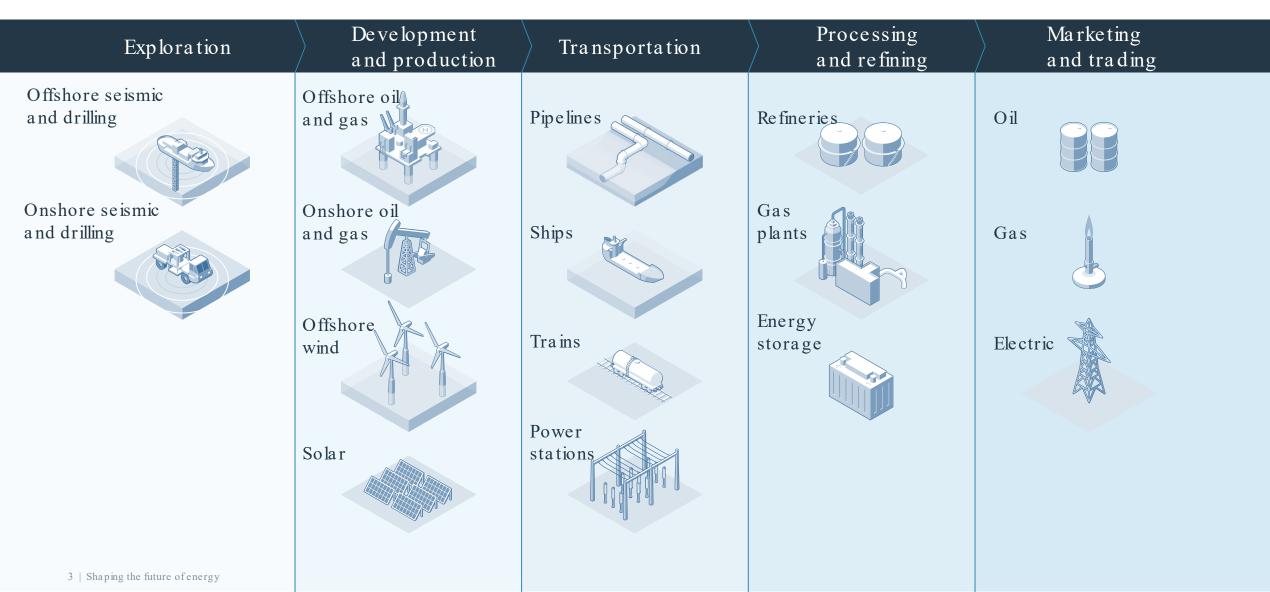


Hydrogen sa fety strategies and risk management in Equinor

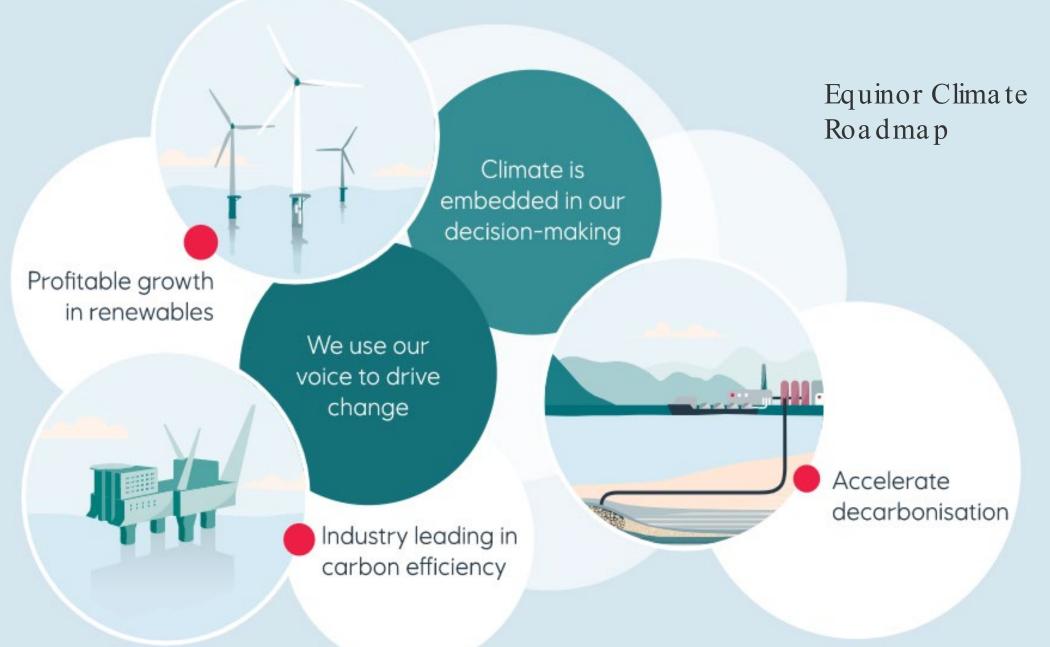
Samdal, U.N. (Presenter), Grainger, D., Hamborg, E.S., Nilsen, S.H., Sommersel, O.K.



Our value chain







Open

Shaping the European future of CCS and clean hydrogen

Competitive edge founded on experience, infrastructure and customers.



15-30 MTPA

CO₂ transport and storage capacity by 2035

Equinor share

>25%

CO₂ transport and storage market share in Europe by 2035 3-5 MAJOR INDUSTRIAL CLUSTERS

Clean hydrogen projects by 2035

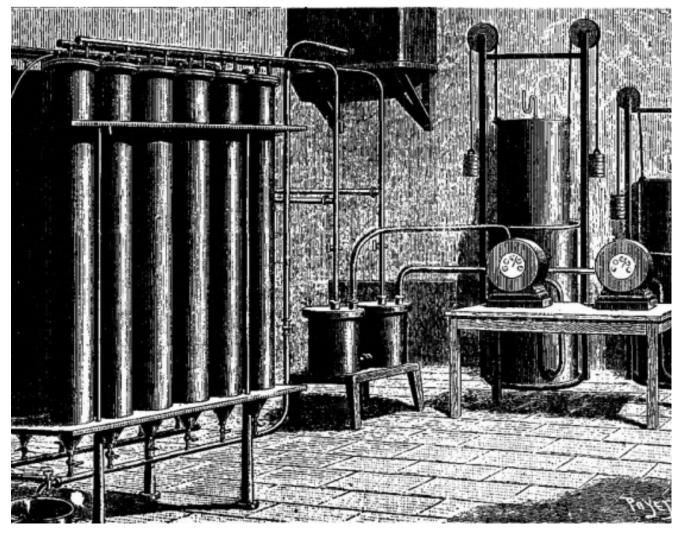
>10%

Clean hydrogen market share in Europe by 2035



Hydrogen utilisation

- Known in industry for more than a century
- Many actors are new to hydrogen, but familiar with hydrocarbons like natural gas (NG)
- Hydrogen is different to natural gas. These differences must be reflected in the design and operation of facilities



Industrial electrolysis of water, early 20th century (https://www.gutenberg.org/files/14990/14990-h/14990-h.htm)

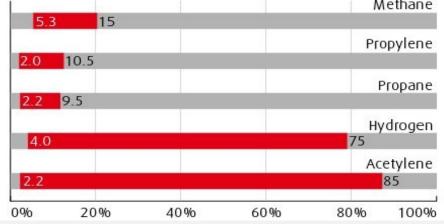


Hydrogen characteristics

Containment (small molecule + failure mechanism) Wear Debris Atomic Hydrogen Diffusion Crack Networks Void Molecular Hydrogen Hydrogen Blistering Hydrogen Embrittlement After penetration, atomic hydrogen Concentration of hydrogen in void increases, pressure reacts to form brittle compounds also increases cracking. and increases cracking.

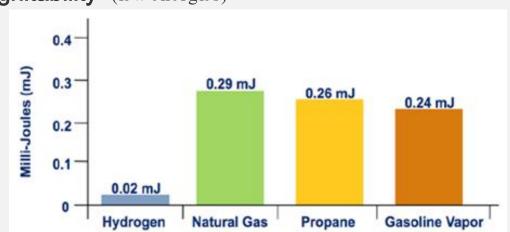
Source: Dini, J.W. (1993) Electrodeposition: The Materials Science of Coatings and Substrates. Noyes Publications

Flammability (flammable concentration in air - wide range) Methane



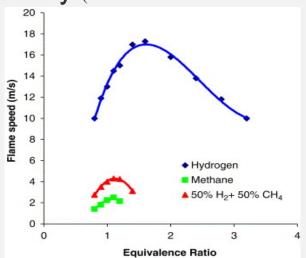
Source: https://www.linde-gas.se/en/safety health ren/gas risks/flammable gas/index.html

Ignitability (low energies)



Source: https://h2tools.org/bestpractices/hydrogen-compared-other-fuels

Reactivity (fast flame acceleration and DDT)

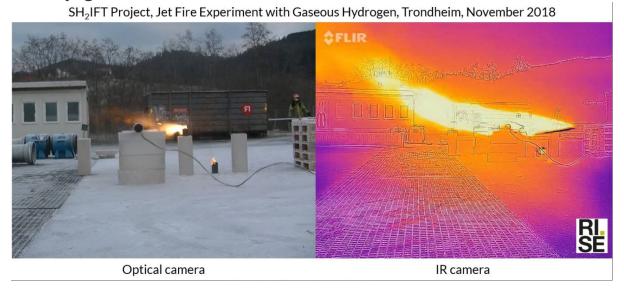


Source: M. Ilbas et.al.: Laminarburning velocities of hydrogen-air and hydrogen-methane-air mixtures: An experimental study. Int. Journal of Hydrogen Energy, vol. 31, issue 12.



Other hydrogen characteristics

- High buoyancy and high diffusivity of gas
- If released in open air the gas will rise and dilute rapidly, unless determined by a jet
 - Note: Evaporation from liquid hydrogen may behave like a heavy gas
- Hydrogen burns with a light blue flame, practically invisible in daylight
- Very low boiling point (20 K, -253 °C.)



https://twitter.com/RISEFR/status/1068457127851761665/photo/1



Two hydrogen accident case studies



Both occurred at Herøya industrial site, Porsgrunn, Norway

- 1. Explosion in an ammonia plant, 1985
- 2. Explosion of a hydrogen-air mixture in a pipeline for CO₂ transportation, 1997



Explosion in ammonia plant, 1985 (presented in ICHS 2005*)

- Two fatalities, extensive damage
- A combination of operational error, technical failures and weakness in the design
- 10-20 kg of hydrogen leaked inside the building
- Most likely that a hot bearing ignited the gas cloud
- 3.5-7 kg of hydrogen involved in the explosion
- From the damage observed, detonation seems most likely
- The explosion caused large number of fragments representing a severe hazard
- Glass windows were broken up to 700 m from the centre of the explosion.
- Within a radius of 100 m all ordinary windows were broken.

 $^{^*}$ Bjerketvedt, Dand Mjaavatten, A $(20\,0\,5)$, A Hydrogen Explosion in a Process Plant - A Case History, Conference paper, ICHS





Explosion of a hydrogen-air mixture in a pipeline, 1997

(Published in Process safety progress, Vol. 20, Issue 1, 2001*)

- No injuries (Sunday a sternoon), but extensive damage
- A combination of operational error, technical failures and weakness in the design (again)
- CO₂ pipeline had been out of service for six days
- Hydrogen leaked into pipeline
- Ineffective purging, air leaking into pipeline
- Possibly ignited by external hot work
- 10 kg of hydrogen involved in explosion
- 850 meters of the pipeline was totally destroyed
- Combustion front accelerated in pipeline, causing rupture at intervals of 1-20 m



Pande, J.O. and Tonheim, J: Ammonia plant NII: Explosion of hydrogen in a pipe line for CO₂. Process safety progress, Vol 20, Issue 1,2001 Pande, J.O., Stokke, R.G., Tonheim, J., Explosion of hydrogen in a pipe line for CO₂, Loss Prevention Bulletin of IChemE 156(1):11-13,2000

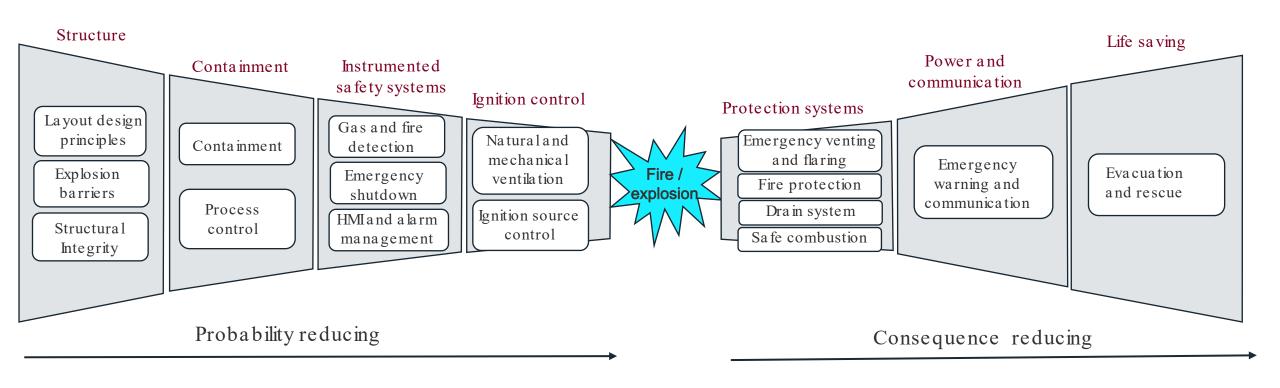


Sa fety strategy and its role in sa fety and risk management





Bow-tie, flammable gas release





Structuralloads

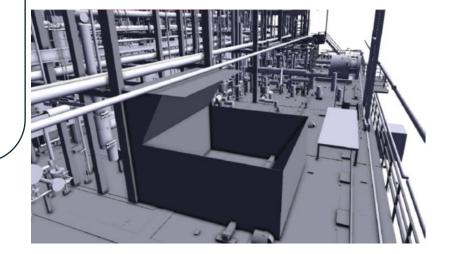
Layout & explosion barriers

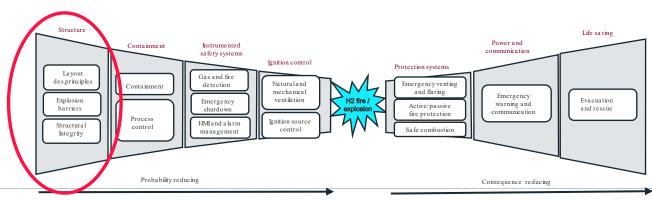
- Avoid enclosed and confined areas
- 'Open box' design to direct leaks away from confined and congested areas
- H₂ piping on top of buildings/structures, outdoors
- Safe location of vent system outlets
- Restricted flow of hydrogen
- Explosion relief panels
- Design for explosion loads
- Design to avoid detonation

Layout design principles

Explosion barriers

Structural Integrity







Ensure containment

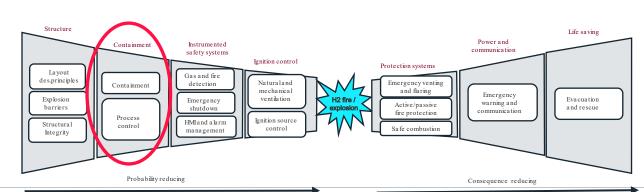
Containment

- H₂-suitable materials of construction
- Fully welded systems
- «Pipe-in-Pipe» solutions
- ESD valves

Process control

Process Control

- Pressure, temperature and level control
- Pressure relief systems
- H₂ vent system





Detection and shutdown

Gas and fire detection

- Specific H₂ gas & flame detectors, sonic detection
- Close to typical leak points
- Early detection in «pipe-in-pipe» in confined areas

Gas and fire detection

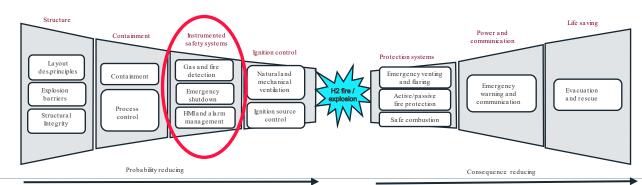
Emergency shutdown

HMI and a larm management

Instrumented safety systems

Emergency shutdown

• Valves automatically to safe position in case of an accident/leak





Prevent ignition

Natural and mechanical ventilation

- H₂-containing equipment and potential leak points located in open, naturally ventilated areas, high level
- High ventilation rate in confined areas

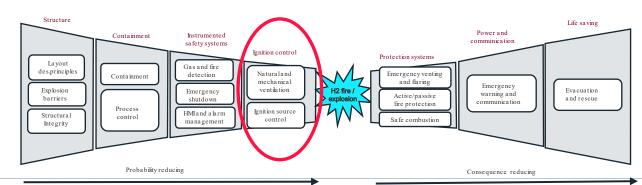
Natural and mechanical ventilation

Ignition source control

Ignition control

Ignition source control

- Equipment Group IIC in H₂ hazardous areas
- Prevent static electricity formation
- Shutdown of ignition sources if confirmed gas detection





Mitigating consequences

Venting and emergency depressurisation

- Vents to be located and designed for ignited scenarios
- Blowdown to safe location
- N₂ purging &pilot flame in flares

Emergency venting and flaring

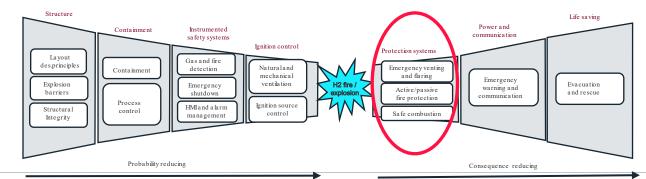
Active/passive fire protection

Safe combustion

Protection systems

Active and passive fire protection

• Protect structures against heat exposure

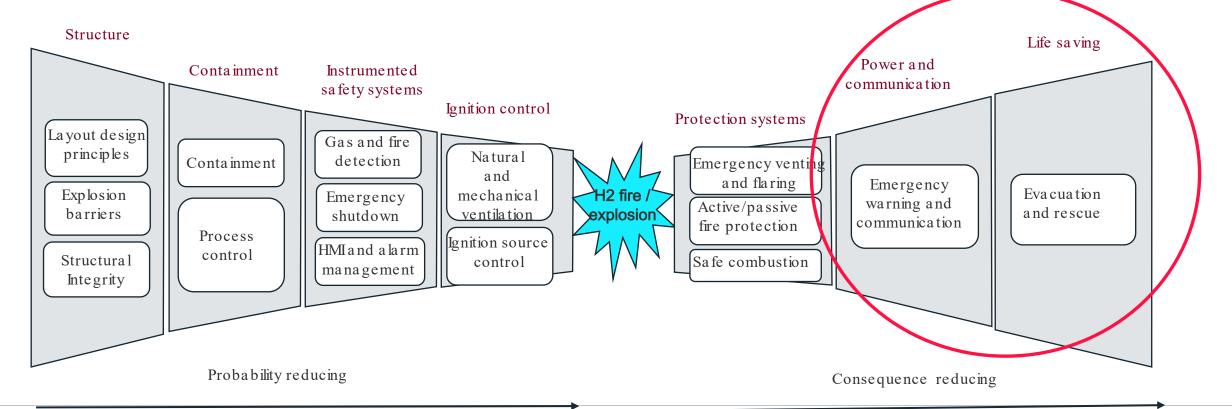




Emergency preparedness and response

• Follow general principles for facilities containing flammable substances

• Emergency response related to *liquid* hydrogen must be considered further





Summing up

- The safety strategy approach used in our oil and gas facilities is valid also for H₂
- Facility design and safety barrier requirements must be adapted to H₂ challenges
- More work needed on safe utilisation of liquid hydrogen
- Sa fety barriers in design, adapted to the properties and risk characteristics of hydrogen, are being implemented in Equinor hydrogen value chain projects

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Thank you!

Hydrogen sa fety strategies and risk management in Equinor

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